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Three new triterpenoid saponins, xuedanglycosides A-C (1–3, resp.), along with six known ones, were isolated from the rhizomes of *Hemsleya chinensis*. By detailed analysis of the NMR spectra, by chemical methods, and by comparison with spectral data of known compounds, the structures of new compounds were determined to be 16a,23a-epoxy- $2\beta,3a,20\beta$ -trihydroxy-10a,23a-cucurbita-5,24-dien-11-on-2-yl β -D-glucopyranoside (1), $2\beta,3a,16a,20\beta$ -tetrahydroxycucurbita-5,25-diene-11,22-dion-2-yl β -D-glucopyranoside (2), and oleanolic acid $28-O-\beta$ -xylopyranosyl-(1 \rightarrow 6)- $O-\beta$ -glucopyranoside (3). In addition, hemslecin A 2- $O-\beta$ -D-glucopyranoside (6), hemsamabilinin B (7), and hemslonin A (9) were obtained for the first time from this plant.

Introduction. - The genus Hemsleya (Cucurbitaceae), containing 31 species, has its centre of distribution in Yunnan and Sichuan Provinces. The tubers of some Hemsleya species are used as folk medicine to treat bronchitis, bacillary dysentery, and tuberculosis [1]. Up to now, more than 80 new triterpenoids and their glycosides have been isolated from this genus [2-16], and some even showed interesting activities. For example, a mixture of hemslecins A (25-acetoxy-23,24-dihydrocucurbitacin F) and B (23,24-dihydrocucurbitacin F) occurring in many *Hemsleya* species is being manufactured in pharmaceutical factories as a treatment for bacterial diseases [17]. Hemslosides Ma2 and Ma3, obtained from Hemsleya chinensis COGN, could increase the water solubility of saikosaponin-a, a pharmacologically active saponin of Bupleuri radix [7]. Cucurbitane compounds are known as bitter principles of many cucurbitaceous plants. However, like several cucurbitane glycosides from the genera Bryonia siraitia, some analogues from the genus *Hemsleya* are sweet tasting [8]. In a previous study, several new compounds were discovered from the tubers of *H. chinensis* [4]. Aimed at finding potentially bioactive and novel compounds, we further investigated this species. As a result, three new triterpene saponins, named xuedanglycosides A - C(1-3), along with six known triterpenoids, hemslecin A (4) [2], hemslecin B (5) [2], hemslecin A 2- $O-\beta$ -D-glucopyranoside (6) [3], hemsamabilinin B (7) [6], oleanolic acid $28 \cdot O \cdot \beta \cdot D$ glucopyranoside (8) [4], and hemslonin A (9) [13] (Fig. 1), were isolated. This report refers to the structural elucidation of the new triterpenoid saponins based on spectroscopic analysis and chemical methods.



Fig. 1. Structures of compounds $1-9^{1}$)

Results and Discussion. - Compound 1 was obtained as a white powder with an optical rotation $[\alpha]_D^{24} = +108.4$ (c = 1.6, MeOH). The molecular formula was determined as $C_{36}H_{56}O_{10}$ by HR-FAB-MS (m/z 647.3798 ($[M-H]^-$; calc. 647.3795), ¹³C-NMR, and DEPT experiments. The IR spectrum (KBr) showed the presence of OH $(3532, 3467, 3371, 3276 \text{ cm}^{-1})$ and C=O (1687 cm^{-1}) groups. After acid hydrolysis of 1 with 3% dry HCl/MeOH, glucose was detected by GC analysis. In the ¹H-NMR spectrum, the aglycone showed resonances for eight Me singlets at $\delta(H)$ 1.14, 1.22, 1.29, 1.36, 1.64, 1.65 (6s, 3 H each) and 1.42 (s, 2×3 H). The anomeric H-atom signal at $\delta(H)$ 5.25 (d, J=7.8) (Table 1) suggested the presence of a β -glucopyranosyl moiety. Comparison of the NMR data of 1 with those of scandenogenin D [10], indicated that the two compounds were very similar except for two Me groups and one additional sugar unit in **1**. Two HO-CH₂ groups at $\delta(C)$ 65.1 (t) and 58.1 (t) due to C(26) and C(27) in scandenogenin D were replaced by two Me groups at δ (C) 18.0 (q, C(26)) and $\delta(C)$ 26.1 (q, C(27)). In addition, C(24) and C(25) were shifted downfield by 0.8 ppm and 8.9 ppm in $\mathbf{1}$, respectively. The chemical shift value of C(2) was shifted downfield from $\delta(C)$ 71.0 (d) in scandenogenin D to $\delta(C)$ 83.5 (d) in **1**, which indicated that the sugar unit was linked to C(2) in **1**. The HMBC correlations from the anomeric H-atom

¹⁾ Arbitrary numbering. For systematic names, see Exper. Part.

	1		2		
	$\delta(\mathrm{H})^{\mathrm{b}})$	$\delta(C)^{c})$	$\delta(\mathrm{H})^{\mathrm{b}})$	$\delta(C)^d)$	
CH ₂ (1)	$1.50 - 1.55 (m, H_a),$	33.4 (<i>t</i>)	1.52-1.57 (overlapped, H _a),	33.3 (<i>t</i>)	
	$2.56-2.58 (m, H_{\beta})$		2.74–2.81 (overlapped, H_{β})		
H-C(2)	4.25 (t, J = 9.2)	83.5 (<i>d</i>)	4.27-4.32 (overlapped)	83.4 (d)	
H-C(3)	3.50 (d, J = 9.1)	80.8(d)	3.53 (d, J = 11.0)	80.8(d)	
C(4)		42.6(s)		42.6(s)	
C(5)		141.8(s)		141.7 (s)	
H-C(6)	5.67 (d, J = 5.0)	119.1 (d)	5.69 (d, J = 3.9)	119.1 (d)	
CH ₂ (7)	1.86 $(t, J = 12.5, H_a),$ 2.20 - 2.27 (m, H_a)	24.3 (<i>t</i>)	1.86 – 1.91 (overlapped, H_a), 2.24 – 2.33 (<i>m</i> , H_a)	24.1 <i>(t)</i>	
$H_{-}C(8)$	1.89 - 1.92 (overlapped)	A2.7(d)	1.86 - 1.91 (overlapped)	43.0(d)	
C(0)	1.05 1.52 (overlapped)	42.7(a)	1.00 1.91 (overlapped)	48.2(s)	
$H_{C(10)}$	2.70 (d I - 12.5)	340(d)	2.74 2.81 (overlapped)	342(3)	
C(11)	2.70(u, j - 12.5)	213.1 (c)	2.74-2.81 (overlapped)	213.2(u)	
$C_{\rm L}^{\rm (11)}$	252(d I = 148 H)	213.1(3)	262 270 (m H)	40.2(3)	
$CH_2(12)$	$2.52(u, J = 14.8, \Pi_{\beta}),$ 2.00(d, L = 14.7, H))	40.0 (1)	$2.02-2.70 (m, \Pi_{\beta}),$	49.5 <i>(l)</i>	
C(12)	$5.00(u, J = 14.7, 11_a)$	49.7 (a)	$5.29 - 5.55 (m, \Pi_a)$	100(a)	
C(13)		40.7(3)		40.0(s)	
C(14)	160 166 (m H)	49.5 (8)	167 171 (availanced)	31.1(3)	
$CH_2(15)$	$1.60 - 1.60 (m, H_a),$	41.8(t)	1.67 - 1.71 (overlapped),	46.4(t)	
II. C(1()	$1.89 - 1.92$ (overlapped, H_{β})	\overline{a}	1.86 – 1.91 (overlapped)	70 4 (1)	
H - C(16)	5.03 - 5.07 (m)	70.6(d)	4.87 - 4.92(m)	70.4(d)	
H - C(1/)	2.13 (d, J = 9.4)	56.3(d)	2.85 - 2.92 (m)	58.9(a)	
Me(18)	1.22(s)	20.2(q)	1.17(s)	20.4(q)	
Me(19)	1.14(s)	20.5(q)	1.16(s)	20.3(q)	
C(20)		72.6(s)		80.0(s)	
Me(21)	1.42 (s)	30.4(q)	1.54(s)	25.3(q)	
CH ₂ (22)	1.73 $(d, J = 13.6, H_a),$ 1.95–1.99 (m, H_{β})	46.4(t)		214.9 (s)	
$H-C(23)$ or $CH_2(23)$	4.90(t, J = 7.5)	71.7 (<i>d</i>)	$3.03-3.13$ (overlapped, H_a), $3.27-3.35$ (<i>m</i> , H_a)	35.8 (<i>t</i>)	
$H-C(24)$ or $CH_{2}(24)$	6.54 (d, J = 8.4)	127.7 (<i>d</i>)	$1.51 - 1.54 (m, H_a),$ 2 53 - 2 58 (m, H_a)	32.3 (<i>t</i>)	
C(25)		133.6(s)	$2.55^{\circ} 2.56^{\circ} (m, \Pi_{\beta})$	145.6(s)	
Me(26) or	1.65(s)	135.0(3) 18.0(a)	4.76(s) 4.83(s)	143.0(3) 110.4(t)	
$CH_2(26)$ Of $CH_2(26)$	1.05 (3)	10.0 (q)	ч.70 (3), ч.03 (3)	110.4 (1)	
Me(27)	1.64 (s)	26.1(q)	1.68(s)	22.9(q)	
Me(28)	1.42 (s)	29.3(q)	1.44 (s)	25.4(q)	
Me(29)	1.29 (s)	22.3(q)	1.33 (s)	22.4(q)	
Me(30)	1.36 (s)	21.1(q)	1.51 (s)	19.1 (q)	
Glc:					
H-C(1')	5.25 (d, J = 7.8)	106.6(d)	5.31 (d, J = 7.6)	106.6(d)	
H-C(2')	4.05 (t, J = 8.5)	76.0(d)	4.04 - 4.10 (m)	76.0(d)	
H-C(3')	4.16(t, J = 8.9)	78.5(d)	4.18 (t, J = 11.0)	78.6(d)	
H-C(4')	4.25 $(t, J = 9.2)$ (overlapped)	71.2(d)	4.27-4.32 (overlapped)	71.2(d)	
H - C(5')	3.81–3.84 <i>(m)</i>	78.6(d)	3.81 - 3.88 (m)	78.6(d)	
CH ₂ (6')	4.23-4.26 (<i>m</i>), 4.33 (<i>dd</i> , <i>J</i> = 11.9, 4.6)	62.4 (<i>t</i>)	4.36 (dd, J = 12.0, 4.4), 4.47 (d, J = 11.6)	62.5 (<i>t</i>)	

Table 1. ¹	H- and ¹³ C-NMR	Data of Compounds	1 and 2 in	$C_5 D_5 N^{\rm a})^1$). δ	in ppm, J in Hz.
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^a) Assignments were established with HSQC, HMBC, and ROESY spectra. ^b) Recorded at 400 MHz. ^c) Recorded at 125 MHz. ^d) Recorded at 100 MHz.

at $\delta(H)$ 5.25 (*d*, H–C(1')) to C(2) ($\delta(C)$ 83.5), as well as from the H-atom signals at $\delta(H)$ 1.65 (*s*, Me(26)) and $\delta(H)$ 1.64 (*s*, Me(27)) to C(24) ($\delta(C)$ 127.7) and C(25) ($\delta(C)$ 133.6) in **1** further confirmed above deduction (*Fig.* 2). The ROESY correlations of H–C(2) to H–C(10)¹) and Me(28), of H–C(3) to Me(29), of H–C(16) to Me(18) and of H–C(23) to Me(21) established the orientations of H–C(2), H–C(3), H–C(16), and H–C(23) as α , β , β , and α , respectively (*Fig.* 3). Hence, the structure of **1** was formulated as $16\alpha,23\alpha$ -epoxy- $2\beta,3\alpha,20\beta$ -trihydroxy- $10\alpha,23\alpha$ -cucurbita-5,24-dien-11-on-2-yl β -D-glucopyranoside.



Fig. 2. Key HMBCs $(H \rightarrow C)$ of compound 1



Fig. 3. Key ROESY (\leftrightarrow) correlations of compound 1

Compound **2** was obtained as a white powder with an optical rotation $[\alpha]_D^{24} = +98.0$ (c = 1.7, MeOH). The molecular formula was determined as $C_{36}H_{56}O_{11}$ from the HR-FAB-MS (negative-ion mode; m/z 663.3726, $[M - H]^-$; calc. 663.3744) and NMR data. The IR spectrum (KBr) indicated OH (3450, 3372 cm⁻¹) and C=O (1689 cm⁻¹) groups. After acid hydrolysis of **2** with 3% dry HCl/MeOH, glucose was detected by GC analysis. The ¹H-NMR spectrum of **2** showed the presence of seven Me *singlets* at $\delta(H)$ 1.16, 1.17, 1.33, 1.44, 1.51, 1.54, 1.68 (7*s*, 3 H each) and a terminal CH₂ group at $\delta(H)$ 4.76 (*s*, 1 H) and 4.83 (*s*, 1 H). The anomeric H-atom signal at $\delta(H)$ 5.31 (*d*, J = 7.6) (*Table 1*) suggested the presence of a β -glucopyranosyl moiety. Careful comparison of the ¹³C-NMR and DEPT data of **2** and hemsamabilinin B (**7**) [6] indicated that the two compounds were very similar, except for the appearance of a C=C bond at $\delta(C)$ 145.6 (*s*, C(25)) and $\delta(C)$ 110.4 (*t*, C(26)) in **2** instead of a quarternary C-atom group at $\delta(C)$ 69.3 (*s*, C(25)) and a Me group at $\delta(C)$ 30.1 (*q*, C(26)) in hemsamabilinin B (**7**). In the HMBC spectrum of **2**, long-range correlations observed from the anomeric H-atom at δ (H) 5.31 (H–C(1')) to C(2) (δ (C) 83.4), as well as from δ (H) 4.76 and 4.83 (*s*, CH₂(26)) to C(25) (δ (C) 145.6) also supported the above suggestion. Therefore, the structure of **2** was elucidated as 2β , 3α , 16α , 20β -tetrahydroxycucurbita-5,25-diene-11,22-dion-2-yl β -D-glucopyranoside.

Compound **3** was obtained as a white powder with an optical rotation $[\alpha]_{D}^{24} = +34.8$ (c = 0.9, MeOH). The molecular formula $C_{41}H_{66}O_{12}$ was deduced from the HR-FAB-MS $(m/z 749.4486 ([M - H]^{-});$ calc. 749.4476), as well as from the ¹³C-NMR and DEPT data. The IR spectrum (KBr) showed absorptions for OH (3533, 3460, 3296 cm⁻¹), C=C (1635 cm⁻¹), C=O (1739 cm⁻¹), and C-O-C groups (1174 cm⁻¹). After acid hydrolysis of 3 with 3% dry HCl/MeOH, glucose and xylose were detected by GC analysis. The ¹³C-NMR signals of two anomeric C-atoms (δ (C) 95.7 (d) and 105.7 (d)), two CH₂ (δ (C) 67.2 (t) and 69.2 (t)), and seven CH groups (δ (C) 70.9–78.8) (Table 2) were consistent with glucose and xylose units. Two anomeric H-atom signals at $\delta(H)$ 6.27 (d, J=8.2) and 4.92 (d, J=7.5) suggested the presence of a β glucopyranosyl and β -xylopyranosyl moiety, respectively. The C-atom signal $\delta(C)$ 62.2 (t) attributable to the C(6') of the sugar in oleanolic acid 28-O- β -D-glucopyranoside (8) [4] was shifted downfield to $\delta(C)$ 69.2 (t, C(6')) in **3**, which indicated that the additional sugar was attached to C(6') of this sugar moiety in **3**. This was confirmed by the HMBC correlation from the anomeric H-atom at $\delta(H)$ 4.92 (d, H–C(1'')) to C(6') ($\delta(C)$ 69.2). Consequently, the structure of **3** was deduced as oleanolic acid $28-O-\beta$ -xylopyranosyl- $(1 \rightarrow 6)$ -*O*- β -glucopyranoside.

The compounds 4-9 were identified by comparison of their spectroscopic data with literature values. Compounds 6, 7, and 9 were obtained for the first time from this plant.

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Experimental Part

General. Glucose and xylose were purchased from Sigma (USA) and New Jersey (USA), resp. Column chromatography (CC): silica gel (SiO₂; 200–300 mesh, Qingdao Marine Chemical, P. R. China); Lichroprep RP-18 (40–63 µm, Merck, Darmstadt, Germany); and Sephadex LH-20 (Pharmacia Fine Chemical Co., Ltd.). Fractions were monitored by TLC, and spots were visualized by heating TLC sprayed with 10% H₂SO₄. GC: Shimadzu GC-17A gas chromatograph equipped with an H₂ flame ionization detector; column: TC-1 capillary column (30 m × 0.25 mm); detector, FID. Optical rotations: JASCO DIP-370 digital polarimeter. IR Spectra: Shimadzu IR-450 instrument; in cm⁻¹; KBr pellets. NMR Spectra: Bruker AV-400, or DRX-500 instruments; chemical shifts (δ) in ppm; TMS as the internal standard; J in Hz. FAB-MS and HR-FAB-MS: VG-AUTOSPEC-3000 spectrometer; in m/z (rel. int. in % of the base peak).

Plant Material. The tubers of *H. chinensis* were collected at Dongchuan County, Kunning City, Yunnan Province of China, in October 2004. It was identified by Prof. S. K. Chen, and a specimen (No. KIB20050623) was deposited with the Laboratory of Phytochemistry, Kunning Institute of Botany.

Extraction and Isolation. The dried and powdered tubers of *H. chinensis* (3.84 kg) were extracted with MeOH (71×6 , each 8 h) at 60°. After removal of the solvent under reduced pressure, a residue (1.21 kg) was obtained. This residue was dissolved in H₂O (31), and then extracted successively with AcOEt (21×3) and BuOH (21×3). The AcOEt and BuOH layers were concentrated to dryness, resp., to give an AcOEt (220.10 g) and a BuOH extract (309.40 g). The AcOEt extract was subjected to CC

Table 2. ¹*H*-and ¹³*C*-*NMR* Data of Compound **3** in $C_5D_5N^a$). δ in ppm, J in Hz.

	$\delta(\mathrm{H})^{\mathrm{b}})$	$\delta(C)^c)$		$\delta(\mathrm{H})^{\mathrm{b}})$	$\delta(C)^c)$
CH ₂ (1)	0.97 (d, J=4.1), 1.52 (dd $J=13.2, 3.3$) (overlapped)	39.0 (<i>t</i>)	CH ₂ (16)	4.95-5.09(m), 5.10-5.14(m)	23.4 (<i>t</i>)
CH ₂ (2)	1.52 (au, y = 15.2, 5.5) (overlapped) 1.78 - 1.82 (overlapped), 2.30 - 2.36 (m)	28.1 (<i>t</i>)	C(17)	5.10-5.1 4 (<i>m</i>)	47.1 (s)
H-C(3)	3.42 (dd, J = 10.6, 5.1)	78.1(d)	H - C(18)	3.19 (dd, J = 13.7, 4.1)	41.8(d)
C(4)		39.4 (s)	CH ₂ (19)	1.24 (d, J = 4.2), 1.70 - 1.75 (m)	46.3 (<i>t</i>)
H-C(5)	0.82 - 0.87 (m)	55.9 (d)	C(20)		30.8(s)
CH ₂ (6)	1.31–1.37 (overlapped), 1.45–1.53 (<i>m</i>)	18.8 (<i>t</i>)	CH ₂ (21)	1.10–1.14 (<i>m</i>), 1.31–1.37 (overlapped)	34.0 <i>(t)</i>
CH ₂ (7)	1.31–1.37 (overlapped), 1.82–1.87 (<i>m</i>)	33.2 <i>(t)</i>	CH ₂ (22)	1.31–1.37 (overlapped), 1.45–1.48 (<i>m</i>)	32.6 (<i>t</i>)
C(8)		39.9 (s)	Me(23)	1.21 (s)	28.8(q)
H-C(9)	1.64 (dd, J = 10.7, 7.1)	48.2(d)	Me(24)	1.02(s)	16.6(q)
C(10)		37.4 (s)	Me(25)	0.93(s)	15.7(q)
CH ₂ (11)	0.83–0.87 (overlapped), 1.86–1.95 (<i>m</i>)	23.9 (<i>t</i>)	Me(26)	1.16 (s)	17.6 (q)
H - C(12)	5.42 (t, J = 3.3)	122.9(d)	Me(27)	1.21(s)	26.2(q)
C(13)		144.2 (s)	C(28)		176.6 (s)
C(14)		42.2 (s)	Me(29)	0.87 (s)	33.2(q)
CH ₂ (15)	1.17 $(d, J = 3.4),$	28.3 (<i>t</i>)	Me(30)	0.87 (s)	23.7 (q)
	1.78 - 1.82 (overlapped)		37.1		
Glc:		055(1)	Xyl:		405 5 (1)
H - C(T)	6.27(d, J = 8.2)	95.7(d)	H-C(1'')	4.92(d, J = 7.5)	105.7(d)
H-C(2')	4.08 - 4.13 (m)	73.9(d)	H - C(2'')	3.98(t, J = 6.0)	74.9(d)
H-C(3')	4.23(t, J = 8.9)	78.8(d)	$H-C(3^{\prime\prime})$	4.08 - 4.13 (m)	77.9(d)
H-C(4')	4.34 – 4.38 (<i>m</i>)	/1.1(d)	$H-C(4^{\prime\prime})$	4.15 - 4.19(m)	70.9(a)
H-C(5')	4.34 - 4.38(m)	78.2 (<i>d</i>)	$CH_2(5'')$	3.59-3.65 (m), 4.30 (dd, J = 11.3, 5.2)	67.2 (<i>t</i>)
CH ₂ (6')	4.36 (d, J = 7.2), 4.71 (d, J = 9.5)	69.2 (<i>t</i>)			

^a) Assignments were established with HSQC and HMBC. ^b) Recorded at 400 MHz. ^c) Recorded at 100 MHz.

 $(SiO_2; CHCl_3/MeOH 1:0, 50:1, 20:1, 10:1, 0:1)$ to yield five fractions (*Fr. 1–5*). *Fr. 2* (31.15 g) was separated by CC (*RP-18*; MeOH/H₂O 55:45) to afford **4** (20.06 g). *Fr. 3* (19.76 g) was chromatographed (SiO₂; petroleum ether/AcOEt 3:1) to give **5** (10.10 g).

The BuOH extract (309.60 g) was subjected to CC (SiO₂; CHCl₃/MeOH 20:1, 10:1, 5:1, 2:1, 0:1) to afford five fractions (*Fr. I–V*). *Fr. II* (18.25 g) was chromatographed (SiO₂; CHCl₃/Me₂CO 9:1 and *RP-18*; MeOH/H₂O 60:40) to yield **1** (15 mg), **2** (12 mg), **6** (500 mg), and **8** (70 mg). Compounds **3** (2.75 g) and **7** (40 mg) were isolated from *Fr. III* (30.46 g) by repeated CC (*RP-18*; MeOH/H₂O 55:45 to 75:25). Similarly, *Fr. IV* (40.65 g) was subjected to CC (SiO₂; CHCl₃/MeOH/H₂O 8:2:0.2 and *Sephadex LH-20*; MeOH/H₂O 90:10) to yield **9** (19 mg).

Acid Hydrolysis of 1–3. Compounds 1–3 (each 2 mg) were treated with 3% HCl in MeOH (5 ml) at 92° for 3 h, resp. 5 ml CHCl₃/H₂O (1:1) were used for extraction. The aq. phase was neutralized with Ag_2CO_3 . The filtrate was concentrated to dryness under reduced pressure.

Sugar Determination in 1–3. Each neutralized hydrolysate of 1–3 was dissolved in 0.6 ml of pyridine, then 0.4 ml hexamethyl disilazane and 0.2 ml trimethylchlorosilane were added successively. The mixture was kept at 60° for 10 min in a water bath. Next, the mixture was centrifuged for 20 min at

 1.0×10^4 rpm. The supernatant was subjected to GC analysis under the following conditions: *Shimadzu GC-17A* gas chromatograph equipped with an H₂ flame ionization detector. Column: *TC-1* capillary column (30 m × 0.25 mm). Column temperature: 200°/260°, programmed increase: 3°/min, carrier gas: N₂ (1 ml/min). Injector and detector temperature: 260°; injection volume: 1 µl; split ratio: 1/50. GC Analysis showed the presence of glucose (t_R 12.04) in **1**–3 and xylose (t_R 11.39) in **3**.

Xuedanglycoside A (=16*a*,23*a*-*Epoxy*-2*β*,3*a*,20*β*-trihydroxy-10*a*,23*a*-cucurbita-5,24-dien-11-on-2-yl *β*-D-Glucopyranoside¹) = (15,25,4R,9*β*,16*a*,23S)-1,20-Dihydroxy-9,10,14-trimethyl-11-oxo-16,23-epoxy-4,9-cyclo-9,10-secocholesta-5,24-dien-2-yl *β*-D-Glucopyranoside; **1**). White amorphous powder. $[a]_{D}^{24}$ = +108.4 (*c* = 1.6, MeOH). IR (KBr): 3532, 3467, 3371, 3276, 2969, 2930, 2881, 2729, 1687, 1454, 1377, 1269, 1209, 1160, 1077, 1031, 636, 465. ¹H-NMR (C₃D₅N, 500 MHz): *Table 1*. ¹³C-NMR (C₃D₅N, 125 Hz): *Table 1*. FAB-MS (neg.): 647 ($[M - H]^-$). HR-FAB-MS (neg.): 647.3798 ($[M - H]^-$, C₃₆H₅₅O₁₀; calc. 647.3795).

Xuedanglycoside B (=2 β ,3 α ,16 α ,20 β -*Tetrahydroxycucurbita*-5,25-*dien*-11,22-*dion*-2-yl β -D-*Glucopyranoside* = (15,25,4**R**,9 β ,16 α)-1,16,20-*Trihydroxy*-9,10,14-*trimethyl*-11,22-*dioxo*-4,9-*cyclo*-9,10*secocholesta*-5,25-*dien*-2-yl β -D-*Glucopyranoside*; **2**). White amorphous powder. [α]₂^D = +98.0 (c = 1.7, MeOH). IR (KBr): 3450, 3372, 2971, 2880, 1689, 1651, 1428, 1387, 1079, 1028. ¹H-NMR (C₅D₅N, 500 MHz): *Table* 1. ¹³C-NMR (C₅D₅N, 100 Hz): *Table* 1. FAB-MS (neg.): 647 ([M – OH]⁻). HR-FAB-MS (neg.): 663.3726 ([M – H]⁻, C₃₆H₃₅O₁₁; calc. 663.3744).

Xuedanglycoside C (= Oleanolic Acid 28-O-β-D-Xylopyranosyl-(1→6)-O-β-D-glucopyranoside = 1-O-[(3β)-3-Hydroxy-28-oxoolean-12-en-28-yl]-6-O-β-D-xylopyranosyl-β-D-glucopyranose; **3**). White amorphous powder. [a]²⁶₂ = +34.8 (c = 0.9, MeOH). IR (KBr): 3972, 3533, 3460, 3296, 2944, 2929, 2863, 1739, 1635, 1463, 1388, 1174, 1074, 1043, 995. ¹H-NMR (C₅D₅N, 400 MHz): *Table* 2. ¹³C-NMR (C₅D₅N, 100 Hz): *Table* 2. FAB-MS (neg.): 750 (M⁻). HR-FAB-MS (neg.): 749.4486 ([M – H]⁻, C₄₁H₆₅O₁₂; calc. 749.4476).

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